

Correction to Chemoselective C-4 Aerobic Oxidation of Catechin Derivatives Catalyzed by the *Trametes villosa* Laccase/1-Hydroxybenzotriazole System: Synthetic and Mechanistic Aspects [*The Journal of Organic Chemistry* **2011**, *76*, 820–832 DOI: 10.1021/jo101886s]. Roberta Bernini,* Fernanda Crisante, Patrizia Gentili,* Fabio Morana, Marco Pierini, and Monica Piras

In Table 2, the reported oxidation potentials of compounds 3, 4, and 7 are incorrect. The corrected oxidation potentials are shown below.

Table 2. Oxidation Potential and Dissociation Energies of C-2–H and C-4–H Bonds of Catechin Derivatives 3, 4, and 7

substrate	E^p (V, vs NHE) ^a	BDE _{C–H} (kJ mol ⁻¹)						
		AM1 ^b		B3LYP/6-31G**//		B3LYP/6-311+G**//		tabulated data ^c
		AM1 ^b	error	AM1 ^b	error	AM1 ^b	error	
3 _{C-2}	1.66	296.6				377.4		
3 _{C-4}		306.7				378.2		
4 _{C-2}	1.64	290.4		366.1		367.4		
4 _{C-4}		306.7		377.0		376.6		
7 _{C-2}	1.72	298.6				386.6		
7 _{C-4}		301.4				381.2		
12		296.2	62.8	361.5	-2.5	358.6	0.4	359.0
13		318.8	54.0	393.3	-20.5	387.4	-14.6	372.8
14		301.2	57.8	360.7	-1.7	355.6	3.3	359.0
15		318.0	60.7	392.5	-13.8	387.0	-8.4	378.6
16 _{C-H1}		297.5	61.5	368.6	-9.6	367.8	-8.8	359.0
16 _{C-H2}		301.7	71.1	383.2	-10.5	379.5	-6.7	372.8
17		301.7				359.8		
18		315.5				387.9		
av error on the calcd BDE _{C–H} data (kJ mol ⁻¹)			61.3		9.8		7.0	

^a [Substrate]: 2 mM, at 500 mV/s in CH₃CN containing Bu₄NF 0.1 M. ^b From equation: BDE_{C–H} = ΔH_f^o(R[•]) + ΔH_f^o(H[•]) – ΔH_f^o(R–H). ^c See ref 43.

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